

# The Effect of Surface Area on the Protein Binding Capacity of Membrane-based Cation-exchange Adsorbers

Julie Robinson, Eboni Hobley  
Heather Chenette, Milagro Marroquin, Scott Husson  
Department of Chemical and Biomolecular Engineering  
CLEMSON UNIVERSITY



## Introduction

- An increased demand for protein therapeutics motivate the need for more efficient downstream processing.
- Protein capture remains one of the slowest and most expensive steps of this process
- The current method of resin-based chromatography has a low mass throughput.
- A need exists for a high throughput technology that can accommodate larger batch sizes.

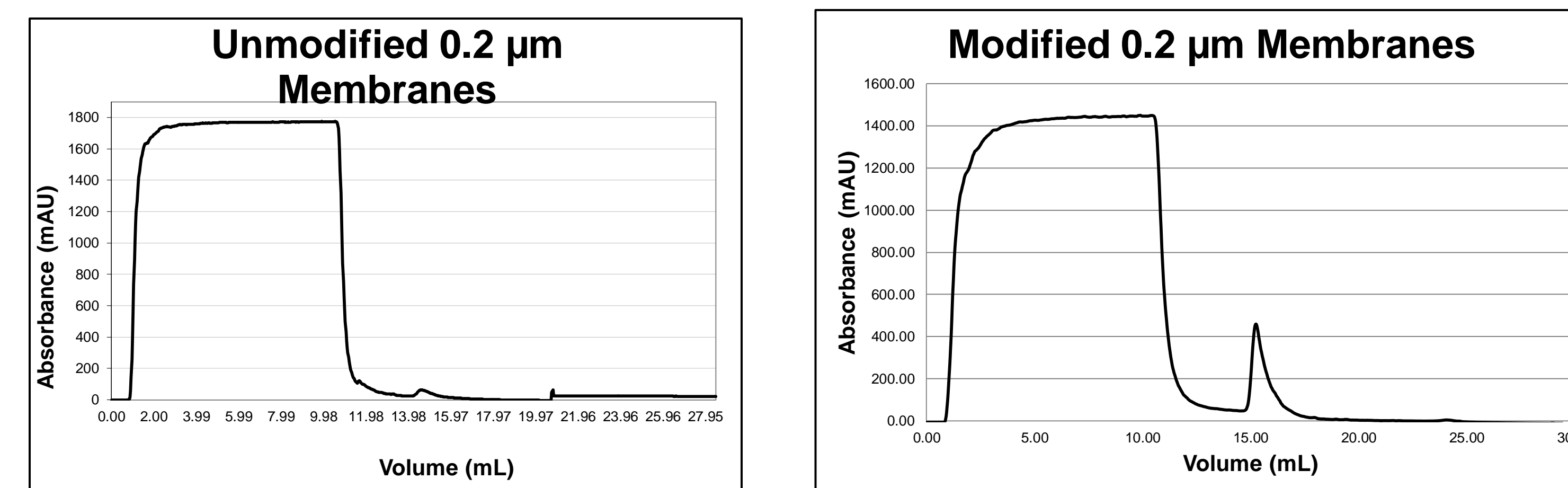
## Objectives

- Surface modify membranes to create cation-exchange adsorbers with high protein binding capacity and high throughput
- Investigate the effect of membrane surface area (pore size) on protein binding capacity of surface-modified membranes
- Compare column performance across a range of flow rates

## Results

- The 0.20  $\mu\text{m}$  pore-size membranes were tested at three different flow rates: 3.85, 7.70, and 12.83 mL/min

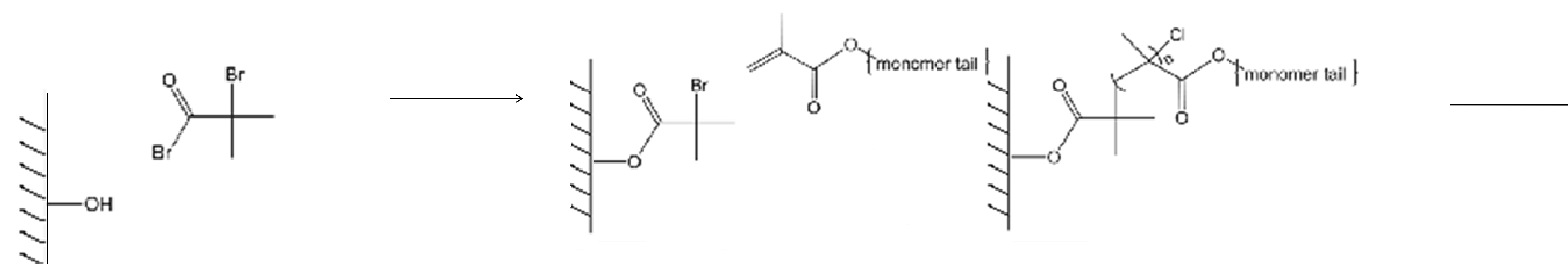
### Typical Chromatograms



### Lysozyme Binding and Elution

Flow Rate (mL/min)	Approximate Dynamic Binding Capacity (mg/mL)	Recovery (%)
3.85	5 ± 2	16.57
7.7	15 ± 1	4.89
12.83	20 ± 3	3.91

## Experimental Methods



### Surface Functionalization

Initiator: 2-BIB (2-Bromoisobutyryl bromide)

### Atom Transfer Radical Polymerization (ATRP)

Monomer: SPMAC (3-sulfopropyl methacrylate, potassium salt)  
Substrate: Membranes of 1.0, 0.45, and 0.20  $\mu\text{m}$  pore sizes  
Solvent: DMSO (Dimethyl Sulfoxide)  
Catalyst: Cu (I)/HMTETA complex  
Monomer Cap: 18C6



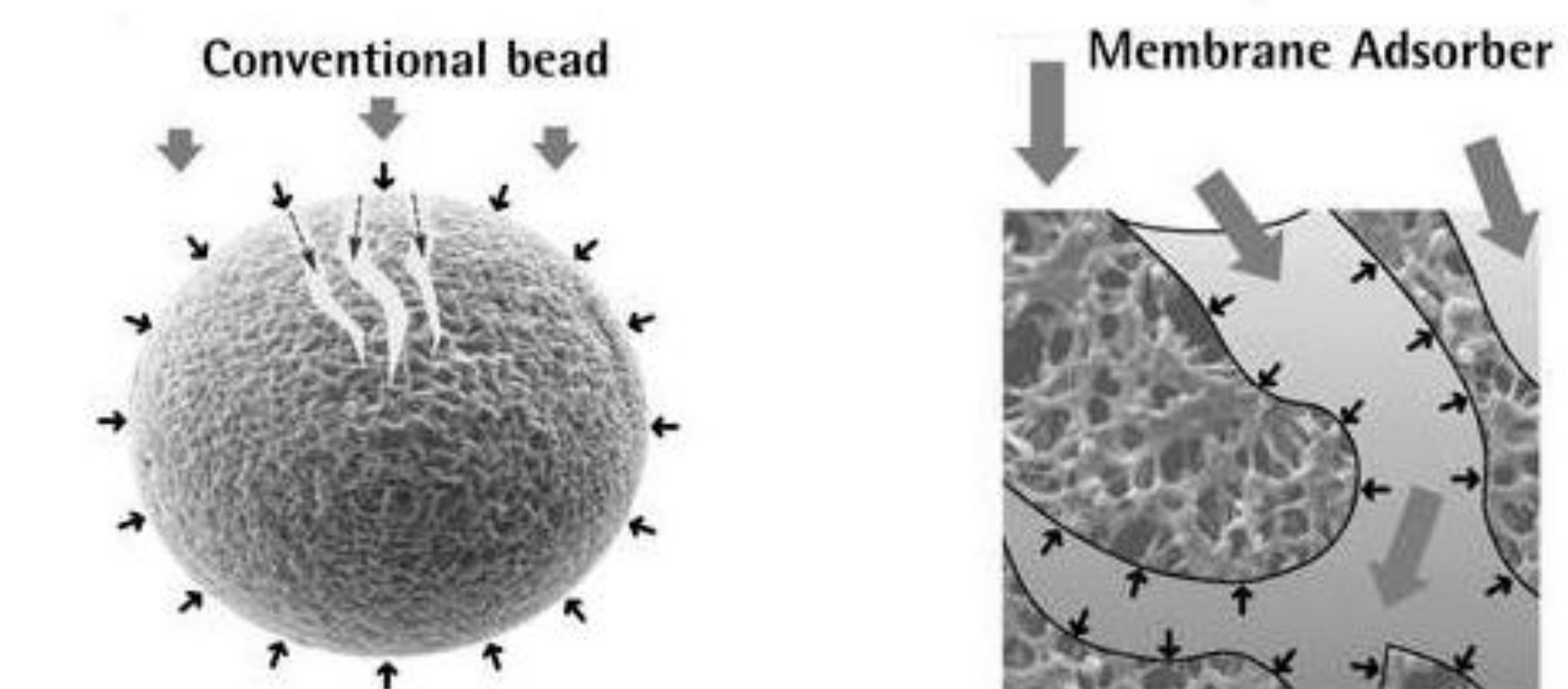
GE Life Sciences

### Dynamic Binding Capacity Measurements

Binding Buffer: 50 mM Tris  
Elution Buffer: 50 mM Tris + 1 M KCl  
Regeneration Buffer: 0.3 M KOH  
Protein: lysozyme in binding buffer

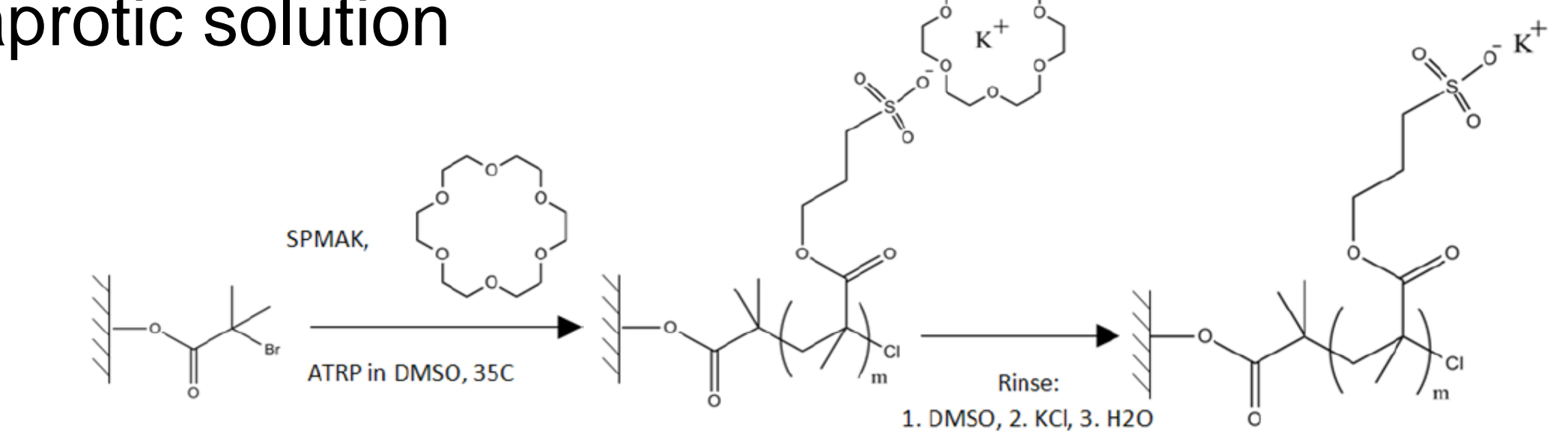
## Discussion

- Membrane chromatography offers lower pressure drop and higher throughput than conventional bead method



Sartorius-Stedim Biotech SA

- Crown ether monomer cap stabilizes acidic monomer in aprotic solution



- Addition of ascorbic acid in 1:1 molar ratio reverses oxidation of Cu (I) catalyst in ATRP solution
- A dry layer thickness of 10 nm was targeted to accommodate 0.20  $\mu\text{m}$  pores. Elipsometry measurements of silicon wafers revealed actual thickness of 5 nm.

$$\text{Approximate Binding Capacity} = \frac{C_0(V_{Br} - V_d)}{V_c}$$

$C_0$  = initial concentration  
 $V_{Br}$  = breakthrough volume (modified)  
 $V_d$  = dead volume (unmodified)  
 $V_c$  = column volume

$$\text{Percent Recovery} = \frac{\text{protein}_{eluted}}{\text{protein}_{bound}} \times 100\%$$

## Conclusions

- The data show how ATRP can be used to create cation-exchange adsorbers and demonstrate the relationship between surface area and dynamic binding capacity

## Future Work

- Investigate the effect of polymer chain grafting density on protein binding capacity of surface-modified membranes
- Determine the optimal polymerization time (chain length) to maximize binding capacity

**Acknowledgements:** This work was part of the Advanced Functional Membranes Research Experiences for Undergraduates program at Clemson University. Support for this REU program was provided by the National Science Foundation under award EEC 1061524.

Visit our website at [www.clemson.edu/ces/chbe/reu/index.html](http://www.clemson.edu/ces/chbe/reu/index.html)

